## **AL Chemistry**

## **Acid-base Equilibrium**

Lewis Acid-base Theory

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An even more general model for acid-base behaviour was suggested by G.N. Lewis in the early 1930s. A **Lewis acid** is an electron pair acceptor, and a **Lewis base** is an electron pair donor. Another way of saying this is that a Lewis acid has an empty atomic orbital that it can use to accept (share) an electron pair from a molecule that has a pair of electrons (Lewis base).

